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## Enantioselective Preparation of 8-Oxabicyclo[3.2.1]octane Derivatives via Asymmetric [3+2]-Cycloaddition of Platinum-Containing Carbonyl Ylides with Vinyl Ethers

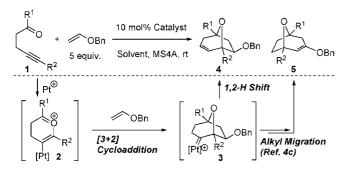
Kento Ishida, Hiroyuki Kusama, and Nobuharu Iwasawa\*

Department of Chemistry, Tokyo Institute of Technology, O-okayama, Meguro-ku, Tokyo 152-8551, Japan

Received March 22, 2010; E-mail: niwasawa@chem.titech.ac.jp

Platinum(II)- or gold(I)-catalyzed electrophilic activation of alkynes has attracted much attention as an efficient method to facilitate atomeconomical construction of complex molecules.<sup>1</sup> Asymmetric versions of these reactions would be highly attractive; however, there are only a few successful reports of asymmetric enyne cyclizations<sup>2</sup> and asymmetric reactions of carbene complexes generated by isomerization of propargyl esters.<sup>3</sup> Herein, we report a platinum-catalyzed enantioselective preparation of synthetically useful 8-oxabicyclo[3.2.1]octane derivatives via an asymmetric [3+2]-cycloaddition reaction of a platinum-containing carbonyl ylide as a new type of asymmetric reaction based on the electrophilic activation of alkynes.

## Scheme 1



Recently, we reported that treatment of acyclic  $\gamma$ , $\delta$ -ynones 1 with a catalytic amount of platinum(II) chloride in the presence of vinyl ethers gave 8-oxabicyclo[3.2.1] octane derivatives  $5^{4c}$  through the novel bifunctional reactive species, platinum-containing carbonyl ylides 2 (Scheme 1).<sup>4-6</sup> The high utility of the product,  $5^{,7}$ prompted us to develop an asymmetric version of this reaction. For this purpose, we first examined the reactivity of platinumphosphine complexes; however, the low electrophilicity of the model complex, cis-PtCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, resulted in recovery of the starting materials. We then thought of using cationic platinum(II)-phosphine complexes<sup>2a-e,8</sup> generated by treatment of platinum(II) chloridephosphine complexes with a silver salt. As expected, the reaction proceeded smoothly at room temperature on treatment of acyclic  $\gamma$ , $\delta$ ynone 1a ( $R^1 = Ph$ ,  $R^2 = Me$ ) with 10 mol % of *cis*-PtCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> associated with 10 mol % AgSbF<sub>6</sub> in the presence of excess benzyl vinyl ether (5 equiv) (Table 1, entry 2), and more importantly, the bicyclic alkene 4a, the product formed through a 1,2-hydrogen shift of the carbene intermediate 3a,<sup>9</sup> was obtained in 45% yield as a single diastereomer, while the hydrolyzed ketone of the product 5a formed by using PtCl<sub>2</sub> was obtained in only 10% yield as a minor product. Furthermore, the reaction with cis-PtCl<sub>2</sub>[P(m-tol)<sub>3</sub>]<sub>2</sub> afforded 4a in higher yield and selectivity (entry 3).<sup>10</sup> Use of 2 equiv (20 mol %) of AgSbF<sub>6</sub> resulted in polymerization of the vinyl ether (entry 4). It should also be noted that the reaction was significantly accelerated compared to the reaction of PtCl<sub>2</sub>. To our knowledge, the electrophilic activation

Table 1. Reaction of 1a ( $R^1 = Ph$ , $R^2 = Me$ ) with Benzyl Vinyl	
Ether in the Presence of Platinum(II)-Phosphine Complexes	

		( )			
entry	Catalyst	Solvent	Time	4a	5a
1	PtCl <sub>2</sub>	toluene	3 days	trace	74%
2	cis-PtCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	$CH_2Cl_2$	3.5 h	45%	$10\%^{a}$
	$AgSbF_{6}$ (1: 1)				
3	cis-PtCl <sub>2</sub> ( $m$ -tol <sub>3</sub> ) <sub>2</sub>	$CH_2Cl_2$	1.5 h	92%	8%
	$AgSbF_{6}$ (1: 1)				
4	cis-PtCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	$CH_2Cl_2$	2 h	16%	—
	$AgSbF_{6}$ (1: 2)				

<sup>a</sup> Obtained as a hydrolyzed ketone.

of alkynes by a monocationic platinum–bisphosphine complex depicted as  $[PtCl(phosphine)_2]^+$  is quite rare,<sup>11</sup> while some enyne cyclizations catalyzed by dicationic platinum–bisphosphine complexes were reported.<sup>2a,b,8a</sup>

Then, we examined various chiral phosphines under the cationic conditions and found that the reaction with (*R*)-BINAP, (*R*)-SEGPHOS, (*S*,*S*)-DIOP, and Josiphos **6** gave the product **4a** with low enantiose-lectivity along with a small amount of the bicyclic enol ether **5a** (ee's were not determined) (Table 2).<sup>12</sup> On the other hand, when Walphos

Table 2. Screening of Chiral Bisphosphine<sup>a</sup>

			4a		5a
entry	Bisphosphine	Time	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>	Yield (%) <sup>d</sup>
$1^e$	(R)-BINAP	7.5 h	21	4 (+)	17
2	(R)-SEGPHOS	25 h	33	18 (+)	detected
$3^e$	(S, S)-DIOP	3 h	67	4 (-)	31
4	Josiphos 6	22 h	50	10(-)	16
5	Walphos 7a	rt-reflux	_	_	_
6	Walphos 7b	20.5 h	31	87 (-)	_
7	Walphos 7c	rt-reflux	trace	_	-
8	Walphos 7d	21.5 h	49	91 (-)	_
$9^e$	Walphos 7d	21.5 h	57	91 (-)	_
$10^{e,f}$	Walphos 7d	16.5 h	70	91 (-)	-
11	Walphos 7e	21.5 h	30	77 (-)	_

<sup>*a*</sup> Reactions were performed by addition of AgSbF<sub>6</sub> (10 mol %) to a mixture of ynone **1a**, benzyl vinyl ether (5 equiv), PtCl<sub>2</sub>(cod) (10 mol %), and bisphosphine (10 mol %) at RT. <sup>*b*</sup> Isolated yield. <sup>*c*</sup> Determined by chiral HPLC analysis (CHIRALPAK AD-H, 0.5 mL/min, 2-propanol/hexane = 1/99) <sup>*d*</sup> NMR yield. <sup>*e*</sup> Isolated PtCl<sub>2</sub>(bisphosphine) complex was used instead of PtCl<sub>2</sub>(cod) and bisphosphine. <sup>*f*</sup> 10 equiv of benzyl vinyl ether were used.

Josiphos : 
$$Ar^{1}_{2}P$$
 Fe :  $PAr^{2}_{2}$   
Fe :  $PAr^{2}_{2}$  6  $Ar^{1} = Ph, Ar^{2} = 3,5-xylyl$   
Walphos :  $PAr^{1}_{2}$  PAr^{2}\_{2} 7a  $Ar^{1} = Ph, Ar^{2} = 3,5-di-trifluoromethylphenyl$   
 $7b Ar^{1} = Ph, Ar^{2} = Ph$   
 $7c Ar^{1} = Ph, Ar^{2} = Ph$   
 $7c Ar^{1} = 2,5-dimethyl-4-methoxyphenyl$   
 $Ar^{2} = 3,5-di-trifluoromethylphenyl$   
 $7d Ar^{1} = Ph, Ar^{2} = 3,5-xylyl$   
 $7d Ar^{1} = Ph, Ar^{2} = 3,5-xylyl$ 

Table 3. Generality of the Reaction

	+ OF 10 equi	R <sup>3</sup> 10 m	nol% PtCl nol% AgS Cl <sub>2</sub> , MS4	bF <sub>6</sub> ►	
entry	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Time	Yield(%) <sup>a</sup> /Ee(%) <sup>b</sup>
1	Ph	Me	Bn	16.5 h	70/91(-)
2	p-Me-C <sub>6</sub> H <sub>4</sub>	Me	Bn	9 h	70/91(-)
3	$p-CF_3-C_6H_4$	Me	Bn	21 h	79/89(-)
4	CH <sub>2</sub> CH <sub>2</sub> Ph	Me	Bn	11 h	80/91(-)
$5^c$	CH <sub>2</sub> CH <sub>2</sub> Ph	Me	TIPS	43 h	68/94(-)
6	<i>i</i> -Pr	Me	Bn	17 h	89/90(-)
$7^d$	<i>i</i> -Pr	Me	PMB	46 h	69/91(-)
8	(CH <sub>2</sub> ) <sub>3</sub> OTIPS	Me	Bn	8 h	83/93(-)
$9^e$	CH <sub>2</sub> CH <sub>2</sub> Ph	Bu	Bn	26 h	50/96(-)
10 <sup>f</sup>	Ph	CH <sub>2</sub> OBn	Bn	9 h	51/97(-)
$11^{c,g}$	Ph	CH=CH <sub>2</sub>	TIPS	48 h	65/97(-)

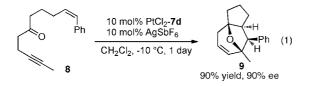
<sup>*a*</sup> Isolated yield. <sup>*b*</sup> Determined by chiral HPLC analysis (see Supporting Information). <sup>*c*</sup> 5 equiv of triisopropylsilyl vinyl ether were used. <sup>*d*</sup> 1.5 equiv of 4-methoxybenzyl vinyl ether and 5 mol % of catalysts were used. <sup>*e*</sup> 5 was obtained as a hydrolyzed ketone in 27% yield. <sup>*f*</sup> Hydrolyzed ketone of 5 and an isomer of 4 were obtained in 17% and 13% yield, respectively. <sup>*g*</sup> Product was isolated as an alcohol by deprotection of the silyl group.

was employed, the product **4a** was obtained with good enantioselectivity as a single diastereomer. Further screening of Walphos ligands revealed that the use of Walphos **7d** gave the product **4a** in 49% yield and 91% ee.<sup>13</sup> Furthermore, the use of the isolated  $PtCl_2-7d$  complex and 10 equiv of vinyl ether increased the yield of **4a** to 70% without lowering the enantioselectivity.

The generality of this asymmetric reaction is summarized in Table 3. Ynones bearing various aryl or alkyl groups as  $R^1$  gave the corresponding products **4** in good yields and mostly in over 90% ee's. 4-Methoxybenzyl vinyl ether could be used as dipolarophiles to give the desired product **4** bearing a PMBO group which can easily be deprotected selectively in the presence of an olefin moiety. Furthermore, triisopropylsilyl vinyl ether could be used as a dipolarophile to give the product **4** with higher enantioselectivity. The reactions of ynones bearing butyl, benzyloxymethyl, and vinyl group as the alkyne substituent  $R^2$  afforded the desired bicyclic alkenes **4** in lower yield but with higher enantioselectivity. In most cases, **4** were obtained as a single diastereomer bearing the alkoxy group in the *exo* orientation.

It should be noted that the products, 8-oxabicyclo[3.2.1]octane derivatives equipped with several functional groups, are useful intermediates not only for the synthesis of related natural products containing this basic skeleton, such as (–)-englerin A<sup>14</sup> and cortistatin,<sup>15</sup> but also for the preparation of a variety of valuable functionalized cyclic compounds through manipulation of the functional groups.

Finally, the reaction was successfully applied to the intramolecular cycloaddition. Thus, treatment of an enynone **8** with 10 mol % of the catalyst gave the desired tricyclic oxacycle **9** in 90% ee in high yield (eq 1).



In summary, we have developed the enantioselective synthesis of potentially useful 8-oxabicyclo[3.2.1]octane derivatives 4 by a simple treatment of acyclic  $\gamma$ , $\delta$ -ynones 1 and vinyl ethers with a

cationic platinum—phosphine complex [PtCl(**7d**)]<sup>+</sup>. To our knowledge, this is the first report of the catalytic enantioselective cycloaddition of metal-containing zwitterionic intermediates generated from alk-4-yn-1-ones.

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**Supporting Information Available:** Preparative methods and spectral and analytical data of all new materials (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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- (10) We believe both electronic and steric parameters of the ligand influence the reaction pathway. Examination of other ligands suggests that bulkier ligands favor a 1,2-hydrogen shift product. Details will be reported in due course.
- (11) For example of the electrophilic activation of alkenes by monocationic platinum-bisphosphine complex, see ref 8c.
- (12) The [3+2]-cycloaddition reaction was thought to proceed in a stepwise manner beginning with the nucleophilic addition of the vinyl ether to the oxonium carbon, which was apart from the chiral ligand on platinum.
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